Understanding The Zhao-Parr Procedure and The Kramers-Henneberger States of Helium

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Certificate of Examination

This is to certify that the dissertation titled **Understanding The Zhao-Parr Procedure and The Kramers-Henneberger States of Helium** submitted by **Mr. Saurabh H. Mhatre(Reg. No. MS13026)** for the partial fulfillment of BS-MS dual degree programme of the Institute, has been examined by the thesis committee duly appointed by the Institute. The committee finds the work done by the candidate satisfactory and recommends that the report be accepted.

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Declaration

The work presented in this dissertation has been carried out by me with Dr. P. Balanarayan at the Indian Institute of Science Education and Research Mohali. This work has not been submitted in part or in full for a degree, a diploma, or a fellowship to any other university or institute. Whenever contributions of others are involved, every effort is made to indicate this clearly, with due acknowledgement of collaborative research and discussions. This thesis is a bonafide record of original work done by me and all sources listed within have been detailed in the bibliography.

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In my capacity as the supervisor of the candidate's project work, I certify that the above statements by the candidate are true to the best of my knowledge.

Dr. P. Balanarayan (Supervisor)

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Saurabh H. Mhatre

"The fundamental laws necessary for the mathematical treatment of a large part of physics and the whole of chemistry are thus completely known, but difficulty lies only in the fact that application of these laws leads to equations that are too complex to be solved."

Paul Dirac

"Without error there can be no brilliancy."

Emanuel Lasker

Contents

Certificate of Examination	iii
Declaration	v
Acknowledgements	vii
List of Figures	xv
List of Tables	xvii
Abbreviations	xix
Physical Constants	xxi

xxv

Abstract

1	Intr	oduction to Density Functional Theory	1
	1.1	Historical background	1
	1.2	Thomas Fermi Model	2
	1.3	1^{st} Hohenberg-Kohn Theorem	3
	1.4	2^{nd} Hohenberg Kohn Theorem	4
	1.5	Levy-Lieb Constrained Search Method	5
	1.6	Kohn Sham Equations	6
	1.7	Momentum Space Densities from Position Space Electron Densities	9
2	Zha	o-Parr Formalism	13
	2.1	Kohn-Sham Equations from Zhao and Parr	13
	2.2	Implementing the self consistent procedure	14
	2.3	Hartree-Fock algorithm	14
		2.3.1 Implementing HF(SCF) method	19
	2.4	Density Fitting Algorithm	20
	2.5	Zhao-Parr KS orbitals Algorithm	21
	2.6	Numerical Results	22

3	Kramer-Henneberger State of He atom	27	
	3.1 Laser-atom interactions	27	
	3.2 Kramer-Henneberger transformation	28	
	3.3 Time Propagation	30	
	3.4 Time Propagation Calculation and Results	30	
	3.5 Future Work	31	
\mathbf{A}	Functional Derivatives	35	
в	3 Zhao-Parr DFT code		
С	Fitting Code	49	

List of Figures

2.1	(a) $\rho(r) = e^{-\pi r^2/4}$ fitting with $\sigma = 2.45 \times 10^{-2} \dots \dots \dots \dots$	21
2.2	Fit of Helium atom ground state density with $\sigma = 2.629 \times 10^{-3}$	21
2.3	(a)Comparing known test density(ρ_0) and calculated density(ρ) (b)Plot	
	of Standard Deviation with varying λ	23
2.4	(a)Comparing known He density(ρ_0) and calculated density(ρ). (Embeddeed plot) calculated wavefunction of He. (b)Plot of Standard De-	
	viation with varying λ .	23
3.1	KH stabilization of helium atom	32

List of Tables

3.1	Gaussian Basis set parameters		31
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Abbreviations

a.u.	atomic units	
KH	Kramers Henneberger	
DFT	D ensity F unctional Theory	
SCF	Self Cnsistent Field	
TDSE	Time Dependent Schrödinger Equation	
KS	${f K}$ ohn ${f S}$ ham	
CI	Configuration Interactions	
\mathbf{HF}	Hartree Fock	

Physical Constants

Speed of Light	c	=	$2.997 \ 924 \ 58 \times 10^8 \ ms^{-1}$
Reduced Planck's constant	\hbar	=	1.054 571 80 × 10 ⁻³⁴ Js
Charge of an electron	e	=	1.602 176 62 × 10 ⁻¹⁹ C
Mass of an electron	m	=	9.109 383 56 \times 10 ⁻³¹ Kg

Dedicated to my loving Grandparents.

Abstract

The first part of the work deals with the implementation of Zhao-Parr procedure to find Kohn-Sham orbitals(Ψ) from a given position space electron densities(ρ), which which a route to obtain momentum space densities(γ). An algorithm for the Zhao-Parr procedure is developed on the basis of Hartree-Fock algorithm. In the second part of the work, helium atom in an intense laser field is analyzed with help of the Kramers-Henneberger framework. The possibility of stabilizing KH state using low frequencies is proposed and being investigated.

Chapter 1

Introduction to Density Functional Theory

According to the quantum mechanical postulate, every physical system consisting of particles is defined by a wave function, which determines every property of the system. Most of the early electronic theory calculations were directed towards determining these wave functions. But these calculations tend to get very complex and tedious with exponential computational complexity for many particles systems. Density functional theory provides us with an alternate approach in which electron density distribution is the focal point rather than the electronic wave function.

1.1 Historical background

Since the inception of wave mechanics, when Schrödinger gave his wave equation in 1926[2], electronic structure calculations were all about approximating wave function of many particle system. Although the most primitive version of DFT-'Thomas Fermi Model' was published only few year after the wave equation, its failure for systems other than jellium model[3] reduced its appeal to the scientific community. About five decades ago, Hohenberg-Kohn(1964)[4] and Kohn-Sham(1965)[5] laid the foundation of Density functional theory, potentially bringing down the complexity of N-electron

systems calculations and cost of computations. It's popularity had gone up exponentially, since 1990's due to the advancement of more accurate approximations of exchange-correlation energy functionals.

1.2 Thomas Fermi Model

Just a few years after Schrödinger gave his wave equation, Lewellyn Thomas and Enrico Fermi separately worked out one of the earliest schemes for solving the many electron problem based on electron density($\rho(\vec{r})$) as a basic variable. The electron density is defined as[6],

$$\rho(\vec{r}) = \int \Psi^*(\vec{r}, \vec{r_2}, \vec{r_3}, ..., \vec{r_N}) \Psi(\vec{r}, \vec{r_2}, \vec{r_3}, ..., \vec{r_N}) dr_2 dr_3 ... dr_N$$
(1.1)

In this model, the energy equation is given in terms of a functional of electron density as,

$$E^{TF}[\rho(\vec{r})] = T^{TF}[\rho(\vec{r})] + \int \rho(\vec{r}) v_{ext}(\vec{r}) d^3r + \frac{1}{2} \int \int \frac{\rho(\vec{r})\rho(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3r d^3r'$$
(1.2)

The last two terms are nuclear-electron Coulomb attraction $(v_{ext} = -\int Z/|\vec{r} - \vec{R}|d^3R)$ and electron-electron repulsion respectively. Assuming that electrons are uniformly distributed over the space(uniform electron gas), one can obtain using Fermi-Dirac Statistics the first term in which kinetic energy of electrons is given by[3]

$$T^{TF}[\rho(\vec{r})] = C_k \int \rho(\vec{r})^{5/3} d^3r$$
(1.3)

where $C_k = \frac{3}{10}(3\pi^2)^{2/3}$. THe Thomas-Fermi model has many shortcomings. Mainly due to the manner in which kinetic energy is approximated. Electron density distribution is assumed to be uniform but this is not true for real systems. It does not lead to any chemical binding. However, it does give a relation between $\rho(\vec{r})$ and external potential $v(\vec{r})$ experienced by an electron. In one way, this idea was the starting point for Walter Kohn's more practical Density Functional Theory(DFT).

1.3 1st Hohenberg-Kohn Theorem

DFT starts with the question, whether the electron density distribution $\rho(\vec{r})$ characterizes the system exactly. The first founding theorem given by Hohenberg and Kohn takes care of this. It states that the ground state density $\rho(\vec{r})$ of a bound system of interacting electrons in some external potential $v(\vec{r})$ determines this potential uniquely There is a very simple proof for this theorem.

Suppose a system of N electrons in the potential $v_1(\vec{r})$ has a non-degenerate ground state density $\rho(\vec{r})$ with corresponding spatial groundstate wave function Ψ_1 and energy E_1 , then

$$E_1 = \langle \Psi_1 | \hat{H}_1 | \Psi_1 \rangle \tag{1.4}$$

$$= \int v_1(\vec{r})\rho(\vec{r})d^3r + \langle \Psi_1 | (\hat{T} + \hat{U}) | \Psi_1 \rangle, \qquad (1.5)$$

where \hat{H}_1 is a total Hamiltonian of a system in the $v_1(\vec{r})$ potential. \hat{T} and \hat{U} are kinetic energy operator and Coulomb interaction operator respectively.

Now, let us assume that there exists another potential $v_2(\vec{r})$ for the same system which is not equal to $v_1 + constant$ with ground state wave function Ψ_2 which also gives rise to the same $\rho(\vec{r})$, such that,

$$E_{2} = \int v_{2}(\vec{r})\rho(\vec{r})d^{3}r + \langle \Psi_{2}|(T+U)|\Psi_{2}\rangle.$$
 (1.6)

By the variational principle,

$$E_1 < \langle \Psi_2 | H_1 | \Psi_2 \rangle = \int v_1(\vec{r}) \rho(\vec{1}) d^3r + \langle \Psi_2 | (T+U) | \Psi_2 \rangle$$
(1.7)

using the relation given by Equation(1.6),

$$E_1 < E_2 + \int [v_1(\vec{r}) - v_2(\vec{r})]\rho(\vec{r})d^3r$$
(1.8)

Similarly, we have

$$E_2 < E_1 + \int [v_2(\vec{r}) - v_1(\vec{r})]\rho(\vec{r})d^3r$$
(1.9)

Adding Equations (1.8) and (1.9) gives,

$$E_1 + E_2 < E_2 + E_1 \tag{1.10}$$

which is a contradiction. Thus, by reductio ad absurdum the assumption that there exists another potential which gives the same $\rho(\vec{r})$, is invalid. It is worth noting that even though we have proved each potential $(v(\vec{r}))$ is unique to the corresponding density $(\rho(\vec{r}))$, this doesn't assure that each density can represent some kind of potential. This known as v-representability of problem of DFT.

As the density determines the external potential $v(\vec{r})$ and also number of electrons, $N = \int \rho r dr$, it gives us the total Hamiltonian of the system, defining its all properties including the wave function.

1.4 2nd Hohenberg Kohn Theorem

The second theorem given by Hohenberg and Kohn lays down a principle for obtaining solution variationally which is the common procedure in quantum mechanics. It basically states that the exact ground state density minimizes the total energy of a system. In other words, one can find ground state density variationally.

As mentioned earlier, the electron density $\rho(\vec{r})$ determines the external potential $(v(\vec{r}))$ of a system. This $v(\vec{r})$ and the number of electrons(N) determines the total Hamiltonian (\hat{H}) of the system, thus the entire system, including the wave function (Ψ) of the system. If for a certain ground state density (ρ_g) of a system, Ψ_g is the ground state wave function such that,

$$E_g[\rho] = \langle \Psi_g | \hat{H} | \Psi_g \rangle \tag{1.11}$$

then,

$$E_{g}[\rho] = \int v(\vec{r})\rho_{g}(\vec{r})d^{3}r + F[\rho_{g}]$$
(1.12)

where, $F[\rho] = \langle \Psi | \hat{T} + \hat{V} | \Psi \rangle$ is an *universal functional* as for the kinetic energy and interactions energy remain same for N number of electrons and are independent of external potential $v(\vec{r})$.

Now, if we have some $\Psi(\text{not necessarily } \Psi_g)$ which gives some density(again not necessarily ρ_g), then according to the variational principle, energy obtained from this wave functions will be greater than or equal to ground state energy,

$$E_{g}[\rho] \leq E[\rho] = \langle \Psi | \hat{H} | \Psi \rangle$$

=
$$\int \rho(\vec{r}) v(\vec{r}) d^{3}r + F[\rho] \qquad (1.13)$$

such that, minimizing the energy with respect to both Ψ and ρ one gets,

$$E_{g}[\rho] = \int \rho_{g}(\vec{r})v(\vec{r})d^{3}r + \langle \Psi_{g}|\hat{T} + \hat{V}|\Psi_{g}\rangle$$

$$\leq \int \rho^{min}(\vec{r})v(\vec{r})d^{3}r + \langle \Psi^{min}|\hat{T} + \hat{V}|\Psi^{min}\rangle = E[\rho] \qquad (1.14)$$

This is known as second Hohenberg Kohn Theorem or Hohenberg Kohn variational principle. The above proof was given in independent works by Levy[8] and Lieb[9] which is famously known as Levy-Lieb constrained search method.

1.5 Levy-Lieb Constrained Search Method

Following the HK variational principle, $\text{Levy}^{[8]}$ and $\text{Lieb}^{[9]}$ proposed a method for obtaining ground state energy in two steps. Each step consists of minimizing energy by fixing one of the wave function Ψ or density ρ .

In first step, trial density $\rho^*(\vec{r})$ is fixed and trial wave functions Ψ^i which give trial density $\rho^*(\vec{r})$ are varied to find constrained minimum energy.

$$E_v[\rho^{\star}] = min_{\Psi \to \rho^{\star}} \langle \Psi^i | \hat{H} | \Psi^i \rangle$$
(1.15)

$$= \int \rho^{\star}(\vec{r})v(\vec{r})d^{3}r + F[\rho^{\star}]$$
 (1.16)

where,

$$F[\rho^{\star}] = min_{\Psi \to \rho^{\star}} \langle \Psi^{i} | \hat{T} + \hat{V} | \Psi^{i} \rangle$$
(1.17)

In this step, one needs to somehow get or approximate the universal functional and obtain the minimum energy $(E_v[\rho^*])$ for the particular density (or potential). Then in the second step, energy is minimized over all trial densities $\rho^*(\vec{r})$,

$$E_g \le E[\rho^*] = \min_{\rho^*} E_v[\rho^*] \tag{1.18}$$

In this way, the problem is reduced from finding minimum energy with respect to trial wave functions with 3N degrees of freedom to a simpler problem of finding minimum energy with respect to density in 3 dimensional space. Here, one must note that trying to solve $F[\rho]$ explicitly, makes the problem 3N-dimensional. So, the better formulation and approximation of $F[\rho]$ has been the foremost concern among DFT developers, such that problem of ground state energies can be strictly formulated in terms of densities.

1.6 Kohn Sham Equations

In 1928, Hartree proposed an approximate description of the electronic structure of an atom with self-consistent single particle Schrödinger equations, treating every electron moving in an effective single particle potential. So, i^{th} particle equation is given as,

$$\left\{-\frac{1}{2}\nabla_i^2 + v_{eff}(\vec{r})\right\}\psi_i = \epsilon_i\psi_i \tag{1.19}$$

where,

$$v_{eff} = -\frac{Z}{r} + \int \frac{\rho(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3r'$$
(1.20)

Here the first term is an electrostatic attraction between the nucleus and electron with Z as the atomic number. The second term is the potential experienced by an electron due to the continuous electronic density distribution $\rho(\vec{r})$ in the space. The above equations are called as self-consistent Hartree equations.

So, in light of self-consistent Hartree equations, Walter Kohn and Lu Sham laid out self consistent scheme for DFT with the use of HK theorems in 1965. At the start, let's assume a system of noninteracting electrons moving in an external potential $v(\vec{r})$, say nuclear-electron attraction, such that electrons will only possess pure kinetic energy(T_s). Then from the HK variational principle,

$$E_v[\rho] = T_s[\rho(\vec{r})] + \int \rho(\vec{r})v(\vec{r})d^3r \ge E_g$$
(1.21)

where,

$$T_s[\rho] = \sum_{i=1}^{N} \langle \psi_i | -\frac{1}{2} \nabla^2 | \psi_i \rangle \qquad (1.22)$$

and $\rho(\vec{r})$ is defined in terms of single particle orbitals as,

$$\rho(\vec{r}) = \sum_{i=1}^{N} |\psi_i(\vec{r})|^2$$
(1.23)

A constraint is applied on the number of electrons, N, such that, $N = \int \rho(\vec{r}) d^3r$. Using Euler-Lagrange equations and minimizing energy with respect to $\rho(\vec{r})$,

$$\delta E_v[\rho(\vec{r})] = \delta T_s[\rho(\vec{r})] + \int \{v(\vec{r}) - \epsilon\} \delta \rho d^3 r = 0$$
(1.24)

where ϵ is an unknown Lagrange multiplier. After functional differentiation with respect to ψ_i^* to find single particle orbital equations, one gets,

$$\left\{-\frac{1}{2}\nabla^2 + v(\vec{r}) - \epsilon_i\right\}\psi_i = 0 \tag{1.25}$$

The ground state energy, which is in principle exact, for a noninteracting electrons system is given by

$$E_g = \sum_{i=1}^{N} \epsilon_i \tag{1.26}$$

Above orbital equations resemble the Hartree equations, and hence can be solved self-consistently. Following a similar procedure, it is easy to find similar Kohn Sham equations for more complex, interacting systems.

Here the derivation of KS equations for an interacting system of electrons is presented as this result will be useful in the work given in the next chapter. Now, the energy equation for an interacting electrons system is given by

$$E_{v}[\rho(\vec{r})] = T_{s}[\rho(\vec{r})] + \int \rho(\vec{r}) v_{ext}(\vec{r}) d^{3}r + \frac{1}{2} \int \frac{\rho(\vec{r})\rho(\vec{r'})}{|\vec{r} - \vec{r'}|} d^{3}r d^{3}r' + E_{xc}[\rho(\vec{r})] \ge E_{g}(1.27)$$

First two terms are familiar noninteracting kinetic energy and external potential interactions respectively. The third term indicates a coulomb interaction between two electrons or in this case average electrostatic potential seen by the particle at the particular point in the space. The last term, E_{xc} is the exchange-correlation energy of which exact form is unknown. Euler-Lagrange equation for N number of electrons is,

$$\delta E_v[\rho(\vec{r})] = \delta T_s[\rho(\vec{r})] + \int \{v_{eff}(\vec{r}) - \epsilon\} \delta \rho d^3 r = 0$$
(1.28)

where,

$$v_{eff}(\vec{r}) = v_{ext}(\vec{r}) + \int \frac{\rho(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3r' + \frac{\delta E_{xc}[\rho(\vec{r})]}{\delta\rho(\vec{r})}$$
(1.29)

Similar to noninteracting system case, here particle is moving in a effective potential $v_{eff}(\vec{r})$ instead of $v(\vec{r})$. Using earlier treatment, we get similar KS equations,

$$\left\{-\frac{1}{2}\nabla^2 + v_{eff}(\vec{r}) - \epsilon_i\right\}\psi_i = 0 \tag{1.30}$$

We will be using this result in next chapter and derive Kohn Sham equations, when the exact physical electron density, experimental or from accurate wave function based calculations, is known. The quality of approximation for $E_{xc}[\rho(\vec{r})]$ dictates the reliability of DFT ground state, and there has lot of effort been made to approximate form of $E_{xc}[\rho(\vec{r})]$ which simple as well as reasonably accurate.

1.7 Momentum Space Densities from Position Space Electron Densities

Electron densities, both momentum space $\gamma(\vec{p})$ and position space $\rho(\vec{r})$, are experimentally measurable quantities complementing to each other in respective space. The wave functions in momentum space $\Phi(\vec{p})$ and position space $\Psi(\vec{r})$ are complementary quantities and related to each other by Fourier transform[3],

$$\Phi(\vec{p_1}, \vec{p_2}, \dots, \vec{p_N}) = \frac{1}{(2\pi)^{3N/2}} \int \Psi(\vec{r_1}, \vec{r_2}, \dots, \vec{r_N}) \\ \times e^{i\sum_{j=1}^N \vec{p_j} \cdot \vec{r_j}} d^3 r_1 d^3 r_2 \dots d^3 r_N$$
(1.31)

The respective densities are given by[3],

$$\rho(\vec{r}) = N \int |\Psi(\vec{r}, \vec{r_2}, \dots, \vec{r_N})|^2 d^3 r_2 \dots d^3 r_N$$
(1.32)

$$\gamma(\vec{p}) = N \int |\Phi(\vec{p}, \vec{p_2}, \dots, \vec{p_N})|^2 d^3 p_2 \dots d^3 p_N$$
(1.33)

But there is no direct relation between $\rho(\vec{r})$ and $\gamma(\vec{r})$. So the question arises whether one can calculate momentum space density given only the electron density? One can employ Zhao-Parr procedure, which is discussed in detail in next chapter, to obtain Ψ from ρ and then from this Ψ , γ can be easily calculated.
References

- [1] Walter Kohn, Nobel lecture, *Reviews of Modern Physics*, **71**,5(1999).
- [2] Erwin Schrödinger, Ann. Phys. (Leipzig), **79**, 361(1926).
- [3] L. H. Thomas, Proc. Cambridge Philos. Soc. 23, 542(1927).
 E. Fermi, Atti. Accad. Naz. Lincei, Cl. Sci. Fis. MAt. Nat. Rend.6, 602(1927).
- [4] P. Hohenberg and W. Kohn, *Phys. Rev.***136**, B864(1964).
- [5] W. Kohn and L. J. Sham, *Phys. Rev.***140**, A1133(1965).
- [6] A. Szabo and N. S. Ostlund, Modern Quantum Chemistry : Introduction to Advanced Electronic Structure Theory
- [7] J. Harriman, *Phys. Rev. A*, **24**, 2, 680(1981).
- [8] M. Levy, *Phys. Rev. A*, **26**, 1200(1982).
- [9] E. Lieb, in *Physics as Natural Philosophy: Essay in Honor of Laszlo Tisza on His 75th Birthday*, edited by A. Shimony and H. Feshbach(MIT, Cambridge, MA) p. 111 (1982).
- [10] R. G. Parr and W. Yang, Density Functional Theory of Atoms and Molecules(Oxford University Press, Oxford)(1989)
- [11] D. R. Hartree, Proc. Cambridge Philos. Soc. 24, 89(1928).
- [12] Q. Zhao and R. G. Parr, J. Chem. Phys. 98(1),543(1993).
- [13] M. K. Harbola, R. R. Zope, A. Kshirsagar and R. K. Pathak, J. Chem. Phy. 122, 204110(2005).

Chapter 2

Zhao-Parr Formalism

In 1992, Q. Zhao and R. Parr[1] gave a method to calculate Kohn-Sham orbitals from a given(experimental or calculated) electron density. This scheme uses the fact that, density dectates the unique potential for the system(in other words, density defines the system), as stated in first Hohenberg-Kohn theorem.

2.1 Kohn-Sham Equations from Zhao and Parr

In their novel approach Zhao and Parr used the following constraint [1], [2]:

$$C = \int \frac{[\rho(\vec{r}) - \rho_0(\vec{r})][\rho(\vec{r'}) - \rho_0(\vec{r'})]}{|\vec{r} - \vec{r'}|} d^3r d^3r'$$
(2.1)

such that C = 0, when calculated density is equal to target density. The constraint is physically follows from electrostatics[9]. If we apply this constraint, we get Kohn-Sham equation as follows[1], [2], [3],

$$\left\{\frac{1}{2}\nabla^2 + \lambda \int \frac{\rho(\vec{r'}) - \rho_0(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3r'\right\} \psi_i = \epsilon_i \psi_i \tag{2.2}$$

here, $v_{eff}(\vec{r}) = \int \frac{\rho(\vec{r}) - \rho_0(\vec{r})}{|\vec{r} - \vec{r'}|} d^3 r'$ is an external potential of the system and λ is an unknown Lagrange multiplier. So, to get useful Kohn-Sham orbitals out of these equations we need find the value of λ for which $\rho(\vec{r}) \approx \rho_0(\vec{r})$ where $\rho(\vec{r}) = \sum_i |\psi_i(\vec{r})|^2$.

2.2 Implementing the self consistent procedure

Solving Kohn-Sham equations is a self-consistent procedure, as one starts with some trial density $\rho(\vec{r})$, constructs $v_{eff}(\vec{r})$ from it, then solves the equation for wave functions $\psi_i(\vec{r})$; from these wave functions one recalculates the density. This recalculated density should be equal to the density, one started with. If that is not the case, then whole procedure should be repeated with a newly calculated density. It is very normal in electronic structure theory to solve the problems using this procedure. This was first introduced for solving self-consistent Hartree equations. The intricacies of the procedure are detailed in subsequent sections

2.3 Hartree-Fock algorithm

The Hartree-Fock method is a numerical way of solving many body Schrödinger equations for an atom or a molecule described in the Born-Oppenheimer approximation. In this frame work, a single Slater determinant is used to describe energy eigenfunction.

$$\widehat{H}|\Psi_0\rangle = E|\Psi_0\rangle \tag{2.3}$$

The Slater determinant represents the antisymmetric nature of one electron orbitals(wave functions) obeying Pauli exclusion principle. the wave function is given as[6],

$$|\Psi_0\rangle = |\chi_1\chi_2\chi_3...\chi_N\rangle \tag{2.4}$$

where, χ_i is a spin orbital[6],

$$\chi_i(x) = \begin{cases} \psi_j(r)\alpha(\omega) \\ \\ \psi_j(r)\beta(\omega). \end{cases}$$

Hamiltonian is designed within a mean field approximation as,

$$\left\{\hat{h}(1) + \sum_{b \neq a} \hat{J}_b(1) - \sum_{b \neq a} \hat{K}_b(1)\right\} \chi_a(1) = \epsilon_a \chi_a(1)$$
(2.5)

$$\hat{f}(1)\chi_a(1) = \epsilon_a \chi_a(1) \tag{2.6}$$

where,

$$\hat{h}(1) = -\frac{1}{2}\nabla_1^2 - \sum_A \frac{Z_A}{|r_1 - R_A|}$$
(2.7)

is the kinetic energy and the potential energy due to nuclear attraction of electron one. And it is one electron operator. The other two terms are^[6]

$$\hat{J}_b(1) = \int \chi_b^*(2) r_{12}^{-1} \chi_b(2) d^3 r_2$$
(2.8)

which is the Coulomb electron-electron repulsion operator and

$$\hat{K}_b(1) = \int \chi_b^*(2) r_{12}^{-1} \chi_a(2) d^3 r_2$$
(2.9)

is the exchange operator. Both of these operators are two electron operators. And the total Hamiltonian operator $\hat{f}(1)$ is called the Fock operator.

Here, for simplicity, we are only going to consider closed-shell restricted HF formalism. In this formalism, one conveniently integrates out spin part and eigenvalue equations is represented in terms of doubly occupied spatial orbitals ($\psi_a | a = 1, 2, ..., N/2$).

$$\hat{f}(1)\psi_i(1) = \epsilon_i\psi_i(1) \tag{2.10}$$

 $\hat{f}(1)$ is a closed-shell Fock operator,

$$\hat{f}(1) = \hat{h}(1) + \sum_{a}^{N/2} 2\hat{J}_{a}(1) - \hat{K}_{a}(1)$$
(2.11)

To solve these equations numerically, the unknown spatial wave function is expressed in terms of a set of M number of known spatial basis set, such that

$$\psi_i = \sum_{u=1}^M C_{ui}\phi_u \tag{2.12}$$

where i = 1, 2, ..., M. If the basis set $\{\phi_i\}$ is complete, then this will be an exact expansion, leading to a numerically exact solution of the problem. To make it complete one has to take very very large basis set $(M \to \infty)$. But this can not be achieved due to practical computational reasons.

Now, the problem of calculating molecular orbitals reduces to the problem of calculating coefficients C_{ui} of the known basis functions in the expansion.

$$f(1)\sum_{v} C_{vi}\phi_v(1) = \epsilon_i \sum_{v} C_{vi}\phi_v(1)$$
(2.13)

After multiplying by $\phi_u^*(1)$ to the left and then integrating, we get a matrix form of the Hartree-Fock equations,

$$\sum_{v} C_{vi} \int \phi_{u}^{*}(1) f(1) \phi_{v}(1) d^{3}r_{1} = \epsilon_{i} \sum_{v} C_{vi} \int \phi_{u}^{*}(1) \phi_{v}(1) d^{3}r_{1} \qquad (2.14)$$

$$\sum_{v} F_{uv} C_{vi} = \epsilon_i \sum_{v} S_{uv} C_{vi}$$
(2.15)

where, S_{uv} and F_{uv} are overlap matrix and Fock matrix respectively.

$$S_{uv} = \int \phi_u^*(1)\phi_v(1)$$
 (2.16)

$$F_{uv} = \int \phi_u^*(1) f(1) \phi_v(1)$$
 (2.17)

These are known as Roothaan equations. For implementing the Hartree-Fock method, one needs explicit form of Fock matrix F which represents Fock operator $\hat{f}(1)$ [6]

$$F_{uv} = \int \phi_u^*(1)\hat{f}(1)\phi_v(1)d^3r_1$$

= $\int \phi_u^*(1)\hat{h}(1)\phi_v(1)d^3r_1 + \sum_a^{N/2} \int \phi_u^*(1)[2\hat{J}_a(1) - \hat{K}_a(1)]\phi_v(1)d^3r_1$
= $H_{uv}^{core} + \sum_a^{N/2} [2(uv|aa) - (ua|av)]$ (2.18)

where,

$$(uv|yz) = \int \phi_u^*(1)\phi_v(1)r_{12}^{-1}\phi_y^*(2)\phi_z(2)d^3r_1d^3r_2$$
(2.19)

and H_{uv}^{core} is a core-Hamiltonian matrix[6],

$$H_{uv}^{core} = \int \phi_u^*(1)\hat{h}(1)\phi_v(1)d^3r_1$$
(2.20)

By expanding $\hat{h}(1)$, one can see that H_{uv}^{core} matrix consist of two kinds of integrals the kinetic energy integrals (T_{uv}) and nuclear attraction integrals $(V_{uv}^{nucl})[6]$,

$$H_{uv}^{core} = T_{uv} + V_{uv}^{nucl} \tag{2.21}$$

where,

$$T_{uv} = \int \phi_u^*(1) \{ -\frac{1}{2} \nabla_1^2 \} \phi_v(1) d^3 r_1$$
(2.22)

$$V_{uv}^{nucl} = \int \phi_u^*(1) \Big\{ -\sum_A \frac{Z_A}{|r_1 - R_A|} \Big\} \phi_v(1) d^3 r_1$$
(2.23)

Now, returning to the solution of Fock matrix,

$$F_{uv} = H_{uv}^{core} + \sum_{a}^{N/2} \sum_{zy} C_{za} C_{ya}^* [2(uv|yz) - (uz|yv)]$$

= $H_{uv}^{core} + \sum_{zy} P_{yz} [(uv|yz) - (uz|yv)/2]$ (2.24)

where, P is a charge density bond order matrix given as,

$$P_{uv} = 2\sum_{a}^{N/2} C_{ua} C_{va}^*$$
(2.25)

So, the final expression for Fock matrix becomes,

$$F_{uv} = H_{uv}^{core} + G_{uv} \tag{2.26}$$

where H_{uv}^{core} is the one electron integral part and G_{uv} is the two electron integral part. The large number of integral evaluations in calculation of G_{uv} matrix is the major difficulty in Hartree Fock calculations.

Once the Fock matrix is specified, one has to convert the Roothaan's equations $(FC = SC\epsilon)$ into usual eigenvalue equations $(FC = C\epsilon)$ which is the case with orthonormal basis set. So, now if we had a non-orthogonal basis, it should be orthogonalized using transformation matrix (X) such that [6],

$$X^{\dagger}SX = I \tag{2.27}$$

Using symmetric orthogonalization, we can get transformation matrix,

$$X \equiv S^{-1/2} = U s^{-1/2} U^{\dagger} \tag{2.28}$$

where, $s = U^{\dagger}SU$. Now, suppose a new coefficient matrix C' such that C = XC' and substituting it into Roothaan's equations gives us transformed Roothan equations[6], [9],

$$FXC' = SXC'\epsilon$$

$$(X^{\dagger}FX)C' = (X^{\dagger}SX)C'\epsilon$$

$$F'C' = C'\epsilon$$
(2.29)

Solution of above eigen value equation can be found by diagonalizing transformed Fock matrix $(F' = X^{\dagger}FX)$. Ground state energy at each iteration in the HF method is given by,

$$E = \frac{1}{2} \sum_{u} \sum_{v} P_{vu} (H_{uv}^{core} + F_{uv})$$
(2.30)

2.3.1 Implementing HF(SCF) method

1. First, specify the atom or molecule ie. nuclear charge(Z_A), coordinates(R_A) and number of electons(N) and choose a known basis set. Taking into consideration computational ease and reasonable reliability, we chose single centered s-type Gaussian basis set.

$$\phi_i(r) = N e^{-\alpha |r|^2} \tag{2.31}$$

where α is an exponent and N is the normalization constant given by,

$$N = \left(\frac{2\alpha}{\pi}\right)^{3/4} \tag{2.32}$$

2. Calculate S, H^{core} and two electron integrals(ab|cd). In our case these have forms as follows[7],

$$S(a,b) = N^2 \left(\frac{\pi}{(a+b)}\right)^{3/2}$$
 (2.33)

$$H^{core}(a,b) = \frac{3ab}{a+b} + \frac{2}{\pi^{1/2}}(a+b)^{1/2}S(a,b)$$
(2.34)

and,

$$(ab|cd) = 2\left(\frac{(a+b)(c+d)}{\pi(a+b+c+d)}\right)^{1/2} S(a,b)S(c,d)$$
(2.35)

3. Get transformation matrix X by diagonalizing S matrix. And get guess P matrix. Here we got P matrix using H^{core} diagonalization.

4. Calculate G matrix from P matrix and the two electron integrals (ab|cd) and obtain Fock matrix $F = H^{core} + G$.

5. Calculate transformed Fock matrix F'.

- 6. Diagonalize F' to get C' and ϵ .
- 7. Get back coefficient matrix C = XC'.

8. Form a new P matrix from coefficient matrix C. If the new density matrix is equal to previous density matrix within some convergence threshold criterion, then we can say that our procedure is converged. And if this is not the case, then repeat the procedure from step 4 with the new density matrix P.

2.4 Density Fitting Algorithm

The Zhao-Parr formalism demands the actual spatial electron density to be provided in order to get the Kohn-Sham orbitals.But this density is often provided in position space grid rather than some analytical form. So, here we propose an fitting algorithm using basis set expansion of Gaussian functions.

Suppose we have been provided with $\rho_0(\vec{r})$ -an grid based density. We can represent it as an expansion in terms of basis set{ φ }.

$$\rho_0(\vec{r}) = \sum_i d_i \varphi_i(\vec{r}) \tag{2.36}$$

Setting up a least squares,

$$\Delta = \min_{d_i} |\rho_0(\vec{r}) - \sum_i d_i \varphi_i|^2 \tag{2.37}$$

After minimizing, we get,

$$\int \rho_0(\vec{r})\varphi_i(\vec{r})d^3r = \sum_j d_j \int \varphi_i^*(\vec{r})\varphi_j(\vec{r})d^3r \qquad (2.38)$$

These are the matrix equations of the form,

$$P_r = S_{ij}D$$
$$D = S^{-1}P_r$$
(2.39)

This is easy enough equation to solve, provided S is invertible. This algorithm is coded and tested for functions and different systems. Some of them are shown below.



ting with $\sigma = 2.45 \times 10^{-2}$.



Zhao-Parr KS orbitals Algorithm 2.5

Given the similarities between Kohn-Sham equations and Hartree-Fock equations, solving them is almost similar to where equations turn into non linear eigenvalue problem, $F^{KS}C = SC\epsilon$. Specifying the Hamiltonian matrix F^{KS} which is analogous to the Fock matrix, is the only difference in between HF method and KS SCF.

In the case of Zhao-Parr formalism, F^{KS} becomes,

$$F_{uv}^{KS} = H_{uv}^{core} + \lambda \int \phi_u^*(\vec{r}) \frac{(\rho(\vec{r'}) - \rho_0(\vec{r'}))}{|\vec{r} - \vec{r'}|} \phi_v(\vec{r}) d^3r d^3r'$$
(2.40)

where, H^{core} solely consist of the kinetic energy [1] and nuclear attraction potential energy part can be included for faster convergence^[3]. So, for single centered s-type Gaussian basis functions, we have already seen H^{core} 's analytical form. The second term in F^{KS} equations, can be broken down into two terms

$$V_{\rho} = \left\langle \phi_u \right| \int \frac{\rho(\vec{r'})}{\left|\vec{r} - \vec{r'}\right|} d^3 r' \left|\phi_v\right\rangle$$
(2.41)

which is nothing but the two electron integral as we solved in HF method and also has analytical form for Gaussian type basis functions as seen earlier. And the other term is,

$$V_{\rho_0} = \left\langle \phi_u \right| \int \frac{\rho_0(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3 r' |\phi_v \right\rangle$$
(2.42)

This term can be solved by numerical integration or by finding an analytical form depending upon the form in which ρ_0 is provided. Using the fitting procedure, discussed earlier, one can represent the grid based density in very convenient Gaussian basis. Then the above equation can be written as,

$$V_{\rho_0} = \left\langle \phi_u | \sum_i \int \frac{d_i \varphi_i(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3 r' | \phi_v \right\rangle$$
(2.43)

And for Gaussian basis($\{\varphi_i\} \equiv e^{-\delta_i r^2}$) this integral takes following form($\{\phi_u\}$ -single centered s-type Gaussian basis set),

$$V_{\rho_0} = N_a N_b N_\delta \frac{2\pi^{5/2}}{(a+b)\delta(a+b+\delta)}$$
(2.44)

After specifying F^{KS} , it is fairly easy task to perform SCF method and obtain Kohn-Sham orbitals.

2.6 Numerical Results

The code is tested for two systems with reasonable results: (1) a test Gaussian density $e^{\pi r^2/4}$ and (2) the helium atom density(He density is obtained from GAMESS). It is observed that as Lagrange multiplier λ increases the density(ρ) tends to converge at given density(ρ_0) till certain critical λ . After that ρ decreases little more and then converges as λ goes to infinity.

Although we would have liked to extend our calculations to the systems with more than two electrons, the increasing complexity and restricted time frame prevented us from doing so. Yet, the ZP procedure remains a promising method and needs more research to be carried out.



FIGURE 2.3: (a)Comparing known test density(ρ_0) and calculated density(ρ) (b)Plot of Standard Deviation with varying λ .



FIGURE 2.4: (a)Comparing known He density(ρ_0) and calculated density(ρ). (Embeddeed plot) calculated wavefunction of He. (b)Plot of Standard Deviation with varying λ .

References

- [1] Q. Zhao and R. G. Parr, J. Chem. Phys. 98(1)(1993).
 Q. Zhao and R. G. Parr, Phys. Rev. A 46, 5, 2337(1992).
- [2] Q. Zhao, R. C. Morrison and R. G. Parr, *Phys. Rev. A* 50, 3, 2138(1994).
- [3] M. K. Harbola, R. R. Zope, A. Kshirsagar and R. K. Pathak, J. Chem. Phy. 122, 204110(2005).
- [4] C. O. Almbladh and A. C. Pedroza, *Phys. Rev. A* **29**, 5, 2322(1984).
- [5] D. R. Hartree, Proc. Cambridge Philos. Soc. 24, 89(1928).

(editors), Elsevier, Amsterdam(2005).

- [6] A. Szabo and N. S. Ostlund, Modern Quantum Chemistry : Introduction to Advanced Electronic Structure Theory.
- [7] Enrico Clementi, MOTECC: Modern Techniques in Computational Chemistry (1989).
- [8] "General Atomic and Molecular Electronic Structure System" M.W.Schmidt, K.K.Baldridge, J.A.Boatz, S.T.Elbert, M.S.Gordon, J.H.Jensen, S.Koseki, N.Matsunaga, K.A.Nguyen, S.Su, T.L.Windus, M.Dupuis, J.A.Montgomery J. Comput. Chem., 14, 1347-1363(1993).
 "Advances in electronic structure theory: GAMESS a decade later" M.S.Gordon, M.W.Schmidt pp. 1167-1189, in "Theory and Applications of Computational Chemistry: the first forty years" C.E.Dykstra, G.Frenking, K.S.Kim, G.E.Scuseria
- [9] C. C. J. Roothan, New development in molecular orbital theory, *Rev. Mod. Phys.* 23, 69(1951).

[10] W. H. E. Schwarz, A. Langenbach and L. Birlenbach, Theor. Chim. Acta. 88, 437-445(1994).

Chapter 3

Kramers-Henneberger State of He atom

3.1 Laser-atom interactions

Laser-atom interactions has been the 'hot topic' over past few decades. According to Einstein's photoelectric effect[1](1905), under electromagnetic radiations, atoms can emit electrons which gain energy from the light photons to get excited and eventually ionize. This causes ionization of the electron in an atom. So, it is very intuitive to say, more the number of photons in the radiation i.e. high intensity, more ionization should occur. And in many cases this is exactly true. But in 1976, Gersten and Mittleman[2] proposed that beyond certain laser conditions electron can get stabilized instead of ionizing due to its quiver motion in a high intensity and high frequency oscillating field.

In the presence of super-intense lasers, when the laser field is comparable or even stronger than normal electric field inside the atom, the system can no longer be treated by traditional perturbation theory. The electron experiences a binding potential due to the nuclear attraction force and the strong laser field instead of plain Coulomb forces which put together an atom. This new system which is stable against ionization is called as laser-dressed atom. As the field gets stronger, the electron, in some sense, becomes free by interacting with light wave and relatively ignoring the nuclear binding forces. But still this unbound electron which has gain massive kinetic energy could not simply break free, because of the oscillating electric force of light wave with electron exhibiting oscillating motion, moving back and forth in the direction of laser polarization. In this process, the 'unbound ' electron becomes stable with respect to ionization.

To fully understand this electron 'stabilization ', one should switch the frame of reference from laboratory frame to oscillating frame of electron where the electron remains stationary and sees a potential as averaged out over its oscillations. This switched frame is popularly known as Kramers-Henneberger frame and the mathematical procedure of switching is called as Kramers-Henneberger transformation.

3.2 Kramers-Henneberger transformation

When KH transformation was first introduce by Henneberger(1968)[3], it was not in the context of laser fields. It was Gavrila *et al.*[4] first used this frame work to explain Hydrogen atom under high frequency laser fields. In this transformation the timedependent Hamiltonian expressed in terms a transformed Hamiltonian in which time dependent part of the Hamiltonian is included into the potential term.

Time-dependent Schrödinger equation in atomic units under dipole approximation for an atom in a oscillating vector field takes the form[5],

$$i\frac{\partial\psi(\vec{r},t)}{\partial t} = \left\{ -\frac{[i\vec{p}+(q/c)\vec{A}(t)]^2}{2m} + V(\vec{r}) \right\} \psi(\vec{r},t)$$
(3.1)

where $\vec{A}(t)$ and $V(\vec{r})$ are the vector and scalar potentials. To achieve the aim of the transformation which is to remove time-dependent dipole term from the Hamiltonian, one can perform unitary transformation on wave function($\psi(\vec{r},t)$),

$$\psi^{KH}(\vec{r},t) = \hat{U}\psi(\vec{r},t) \tag{3.2}$$

where, unitary transform \hat{U} is given as

$$\hat{U} = exp\left\{\frac{i}{\hbar}\int^t \left(\frac{i\hbar q}{mc}\vec{A}(t').\vec{\nabla} + \frac{q^2}{2mc^2}\vec{A}(t')^2\right)dt'\right\}$$
(3.3)

Applying this transformation to the Schrödinger equation gives,

$$i\hbar\frac{\partial}{\partial t}|\hat{U}^{\dagger}\psi^{KH}\rangle = \hat{H}|\hat{U}^{\dagger}\psi^{KH}\rangle \tag{3.4}$$

Simplifying this equation would give,

$$i\hbar \hat{U}^{\dagger} \frac{\partial}{\partial t} \psi^{KH}(\vec{r},t) = \hat{H} \hat{U}^{\dagger} \psi^{KH}(\vec{r},t)$$

$$i\hbar \frac{\partial}{\partial t} \psi^{KH}(\vec{r},t) = [\hat{U}\hat{H}\hat{U}^{\dagger}]\psi^{KH}(\vec{r},t)$$

$$i\hbar \frac{\partial}{\partial t} \psi^{KH}(\vec{r},t) = H^{KH} \psi^{KH}(\vec{r},t)$$
(3.5)

where Hamiltonian is transformed as $\hat{H}^{KH} = \hat{U}\hat{H}\hat{U}^{\dagger}$ which is known as Kramers-Henneberger Transformation.

This transformation expresses the time dependent laboratory frame Hamiltonian into accelerated frame commonly known as Kramers-Henneberger frame.

$$H^{KH} = \frac{\vec{P}^2}{2m} + V(r + \alpha(t))$$
(3.6)

where, $\alpha(t)$ is the displacement of charged particle in its quiver motion.

In the case of electromagnetic field such as laser, where time-dependent electric field vector is given as, $\vec{E} = \alpha_0 \cos(\omega t)$, the transformed Hamiltonian becomes,

$$H^{KH} = \frac{\vec{P}^2}{2m} + V(r + \alpha_0 \cos(\omega t)) \tag{3.7}$$

where $\alpha_0 = \epsilon/\mu\omega^2$ is the amplitude of electric vector and the maximum distance to which electron oscillates.

3.3 Time Propagation

The time dependent Schrödinger equation for electron moving in the electric field is given as[11],

$$i\frac{\partial}{\partial t}\Psi(t) = \{\hat{H} - e\hat{\vec{\mu}}.\vec{E}(t) - i\lambda V_{absorb}\}\Psi(t)$$
(3.8)

where $\hat{\vec{\mu}}$ is dipole operator and V_{absorb} is the absorbing potential which is used to model the ionization. The time evolution of electronic wave function is to be seen. There are different methods which has been employed for solving TDSE. The time propagator method is one of such methods. The general solution of the Schrödinger equation is given in terms of separable form,

$$\Psi(\vec{r},t) = e^{-i\hat{H}t} \Phi(\vec{r}) \tag{3.9}$$

The time dependent term $e^{-i\hat{H}t}$ is treated as the propagator operator. With the help of the Trotter factorization, the time-dependent wave function is propagated[11]

$$\Psi(t + \Delta t) = e^{-i\hat{H}_{el}\Delta t/2}e^{-\lambda\hat{V}_{absorb}\Delta t/2}W^{T}$$

$$\times e^{-iE(t + \Delta t/2)d\Delta t}W$$

$$\times e^{-\lambda\hat{V}_{absorb}\Delta t/2}e^{-i\hat{H}_{el}\Delta t/2}\Psi(t)$$
(3.10)

where, $W^T D W = d$ with D is the transition dipole matrix in the direction of laser field.

3.4 Time Propagation Calculation and Results

Before doing time propagation calculations, full CI calculations of helium atom were performed for Exponentially Tempered Gaussian(ExTG) basis set using GAMESS to generate Configuration Space Functions(CFS) The ExTG basis set is generated using using following expression[12],

$$ln\xi_n = ln\xi_0 + \frac{exp(\beta n) - 1}{\beta} lna_0$$
(3.11)

where ξ is the exponent of a Gaussian function. The basis set is generated using following basis parameters, By CI calculations using above basis set, 4753 CSFs were

Ν	$ln\xi$	$\ln a_0$	β
28	4.939590	-0.506662	-0.03
15	1.741270	-0.301421	-0.03
11	1.844320	-0.291704	-0.03
	N 28 15 11	Ν lnξ 28 4.939590 15 1.741270 11 1.844320	Ν lnξ ln a₀ 28 4.939590 -0.506662 15 1.741270 -0.301421 11 1.844320 -0.291704

TABLE 3.1: Gaussian Basis set parameters

generated.

In our calculations the time-dependent wave function is expanded in the basis of the configuration interactions singles(CIS): $\Psi(t) = \sum_{i=0} C_i(t) |\Psi_i\rangle$ And then using above mentioned time propagation, time-dependent coefficients are solved.

The time propagation calculations are performed for various laser pulse profiles. The stable KH state is found for the pulse with $\alpha_0 = 1.1a.u.$ and $\omega = 0.65a.u.$, at KH energy 1.7548a.u.

The Helium atom KH orbitals are obtained and plotted. Looking at the orbital plots, one can say that under super-intense laser field, electrons of He atom experiences potential resembling to H_2 molecule potential.

3.5 Future Work

Stabilization of the KH state usually demands high frequency high intensity lasers. We would like to explore the possibility of KH stabilization using high intensity but relatively low frequency lasers. Our proposal is, rather than starting from the ground state, to populate a Rydberg state near the KH state, using this Rydberg state as an initial state and then try to find KH stabilization.



FIGURE 3.1: KH stabilization of helium atom

References

- [1] A. Einstein, Ann. Phys. 17, 132(1905).
- [2] J.I. Gersten and M.H. Mittleman, J. Phys. B 9, 2561 (1976).
- [3] W.C. Henneberger, *Phys. Rev. Lett.* 21, 838 (1968).
 H.A. Kramers, *Collected scientific papers* (North-Holland, Amsterdam, 1956).
- [4] M. Gavrila and J. Kaminski, *Phys. Rev. Lett.* 52, 613 (1984).
 M. Pont, N.R. Walet, M. Gavrila and C.W. McCurdy, *Phys. Rev. Lett.* 61, 939 (1988).
 M. Pont, N.R. Walet and M. Gavrila, *Phys. Rev. A* 41, 477 (1990).
- [5] R. M. A. Vivirito and P. L. Knight, *j. Phys. B. At. Mol. Opt. Phys.* 28, 4357-4375(1995).
- [6] F. Morales, M. Richter, S. Patchkovskii and O. Smirnova, PNAS,108, no. 41, 16906-16911(2011).
- [7] J. H. Eberly and K. C. Kulander, *Science*, **262**, 1229(1993).
- [8] John David Jackson, *Classical Electrodynamics* (Wiley, India, 2013).
- [9] D. J. Griffiths, Introduction to Electrodynamics
- [10] "General Atomic and Molecular Electronic Structure System" M.W.Schmidt,
 K.K.Baldridge, J.A.Boatz, S.T.Elbert, M.S.Gordon, J.H.Jensen, S.Koseki,
 N.Matsunaga, K.A.Nguyen, S.Su, T.L.Windus, M.Dupuis, J.A.Montgomery J.
 Comput. Chem., 14, 1347-1363(1993)
 "Advances in electronic structure theory: GAMESS a decade later" M.S.Gordon,

M.W.Schmidt pp. 1167-1189, in "Theory and Applications of Computational Chemistry: the first forty years" C.E.Dykstra, G.Frenking, K.S.Kim, G.E.Scuseria (editors), Elsevier, Amsterdam(2005)

- [11] Q. Liao, Wen Li, and H. B. Schlegel, Can. J. Chem. 94: 989997(2016).
 Paul Hoerner and H. Bernhard Schlegel, J. Phys. Chem. A 121, 1336-1343(2017).
- [12] P. R. Kaprlov-nsk and J. mydke, J. Chem Phys. 138, 024105(2013).
 P. R. Kaprlov-nsk, J. mydke and S. Civi, J. Chem Phys. 139, 104314(2013).

Appendix A

Functional Derivatives

Functional is defined as the multivariable function, $F[f(x_1), f(x_2), ...]$. In other words functional is the function of a function such that F[f(x)]. Derivative of the function is nothing but the variation in function with respect to repective variable. Similar defination applies for for functional derivative. Variation in functional is given as,

$$\delta F[f(x)] = F[f(x+\delta x)] - F[f(x)] \tag{A.1}$$

Derivative for multivariable function is defined as,

$$df = \sum_{i} \frac{\partial f}{\partial x_i} dx_i \tag{A.2}$$

Similarly functional derivative is defined as,

$$\delta F = \int_{a}^{b} \frac{\delta F}{\delta f} \delta f dx \tag{A.3}$$

This defination only holds for finite range ie. for finite integral. One can also interprete the variation in F[f(x)] in more accurate way as,

$$\int \frac{\delta F}{\delta f} \phi(x) dx = \lim_{\epsilon \to 0} \frac{F[f + \epsilon \phi] - F[f]}{\epsilon}$$
(A.4)

where $\phi(x)$ is an arbitrary function. This defination or reasoning is not mathematically rigorous, for furthur information please check 'Theoretical Numerical Analysis: A Functional Analysis Framework by K. Atkinson and W. Han(Springer, Berlin, 2009).

Functional derivatives are at very heart of Lagrangian mechanics, and have been used for solving Euler-Lagrange equations, namely to reduce constraints to potentials. Let us see the simplification for Zhao-Parr constraint.

$$C = \int \frac{(\rho(\vec{r}) - \rho_0(\vec{r}))(\rho(\vec{r\prime}) - \rho_0(\vec{r\prime}))}{|\vec{r} - \vec{r\prime}|} d^3r d^3r \prime$$
(A.5)

By defination,

$$\int \frac{\delta C}{\delta \rho} \phi(\vec{r}) d^3 r = \frac{d}{d\epsilon} \int C[\rho(\vec{r}) + \epsilon \phi(\vec{r})]_{\epsilon=0} d^3 r$$
(A.6)

Simplifying the equation for C,

$$\int \frac{\delta C}{\delta \rho} \phi(\vec{r}) d^3 r = \frac{d}{d\epsilon} \int \frac{(\rho(\vec{r}) - \rho_0(\vec{r}) + \epsilon \phi(\vec{r}))(\rho(\vec{r'}) - \rho_0(\vec{r'}) + \epsilon \phi(\vec{r'}))}{|\vec{r} - \vec{r'}|} d^3 r d^3 r'$$

$$= \int \frac{[\rho(\vec{r}) - \rho_0(\vec{r})]\phi(\vec{r'}) + [\rho(\vec{r'}) - \rho_0(\vec{r'})]\phi(\vec{r})}{|\vec{r} - \vec{r'}|} d^3 r d^3 r'$$

$$= 2 \int \frac{[\rho(\vec{r'}) - \rho_0(\vec{r'})]\phi(\vec{r'})}{|\vec{r} - \vec{r'}|} d^3 r d^3 r'$$
(A.7)

Thus, by comparing both sides, one gets the functional derivative of the C functional,

$$\frac{\delta C}{\delta \rho} = 2 \int \frac{\left[\rho(\vec{r\prime}) - \rho_0(\vec{r\prime})\right]}{\left|\vec{r} - \vec{r\prime}\right|} d^3 r \prime$$
(A.8)

so, v_{eff} 's form has been obtained as given in equation (2.2) with the integral factor is absorbed into the Lagrange multiplier λ

Appendix B

Zhao-Parr DFT code

```
|-----|
MODULE PAR
|-----|
IMPLICIT NONE
integer(kind=4), parameter :: nbas=17, nele=2
END
/_____/
|-----|
program ZP_He
|-----|
USE PAR
implicit none
integer(kind=4), parameter :: npt=1001, nf=17
integer(kind=4) :: i, j, k, l, m, n, o, ip
real(kind=8)
         :: F(nbas,nbas), X(nbas,nbas), TEI(nbas,nbas,nbas,nbas)
real(kind=8)
         :: a(nbas), c(nbas), S(nbas,nbas), H(nbas,nbas), F1(nbas,nbas)
real(kind=8)
         :: expnt(nbas),cof(nbas)
real(kind=8)
         :: V(nbas,nbas), Energy, P(nbas,nbas), su, rh0
          :: core, rep, Energy1, conv, den(npt), check(nbas,nbas)
real(kind=8)
```

```
:: coef(nbas,nbas), coef1(nbas,nbas), ft(nbas,nbas)
real(kind=8)
real(kind=8)
               :: E(nbas), co(nbas),G(nbas,nbas),VO(nbas,nbas), p1(nbas,nbas)
real(kind=8)
                :: xi(npt), y, z, r2, dr, psi(npt), fun, total_density
real(kind=8), parameter :: xmin=0.0000d0, xmax=20.0000d0
real(kind=8), parameter :: pi=3.14159265359d0
real(kind=8)
               :: lambda, std
real(kind=8)
               :: norsum, norsum1, nor1, nor, add, w(npt), trace, rho, del(npt)
real(kind=8) :: ncoef(nbas), nbasis(nbas), amin,amax, dens,ni
call basis(a,c)
call X_matrix(a, c, X)
call Two_electron_integral(a, c, TEI)
call Hcore(a, c, H)
call rho0(a, c, ncoef, nbasis, V0)
call guess_Density_matrix(H, P)
call Hcore(a, c, H)
call overlap(a, c, S)
Energy=0.00d0
Energy1=1.00d0
conv=1.00d0
n=0
open(12, file='lambda_vs_energy')
open(13, file='std1.dat')
open(14, file='lambda_vs_energy.dat')
lambda=10
p1(1,1)=1.00d0
do 1001 m=1, 101
write(*,*) n
do 111 i=1,nbas
  do 112 j=1,nbas
      add=0.00d0
      do 113 k=1,nbas
```

```
do 114 l=1,nbas
          add=add+P(k,1)*(TEI(i,j,1,k)-0.50d0*TEI(i,k,1,j))
          114 enddo
       113 enddo
      G(i,j)=add
!
      F(i,j)=H(i,j)+G(i,j)  !Hartree fock calculations
      F(i,j)=H(i,j)+lambda*(G(i,j)-VO(i,j))
    112 enddo
      write(*,'(7f14.7)') (G(i,j)-VO(i,j), j=1,nbas)
 111 enddo
ft=matmul(F,X)
F1=matmul(transpose(X),ft)
core=0.00d0
do 221 i=1,nbas
   do 222 j=1,nbas
       core=core+P(j,i)*(H(i,j)+F(i,j))
   222 enddo
221 enddo
n=n+1
Energy1 = Energy
Energy=0.5*core
total_density=0.000d0
norsum=0.000d0
norsum1=0.00d0
do 551 i=1, nbas
   do 552 j=1,nbas
      norsum1=norsum1+P1(i,j)*P1(i,j)
      norsum=norsum+P(i,j)*P(i,j)
   552 enddo
551 enddo
nor=dsqrt(norsum)
```

```
nor1=dsqrt(norsum1)
conv=nor1-nor
if (dabs(conv)<1.0e-6)then
   exit
else
   call CALL_DSYEV_WT_EIGVECS(F1, nbas, E, coef1)
   coef=matmul(X,coef1)
  do i=1,nbas
     do j=1,nbas
       p1(i,j)=P(i,j)
       P(i,j)=0.0d0
       do k=1, nele/2
          P(i,j)=P(i,j)+2*coef(i,k)*coef(j,k)
       enddo
     enddo
   enddo
endif
1001 enddo
write(*,*) n, Energy, Energy1, conv, E(1), E(2)
|-----|
! Electron Density
!-----!
call check_constraint(a, p, npt, xmin, xmax, rh0, w, xi, rho)
open(11, file='zphe07w.dat')
dr=(xmax-xmin)/(npt-1)
del=0.00d0
do 331 i=1,npt
  xi=xmin+(i-1)*dr
  r2=(xi(i)*xi(i))
  dens=0.00d0
```

```
do 221 k=1, nbas
     su=0.00d0
     do 441 l=1, nbas
       su=su+coef(1,1)*dexp(-a(1)*r2)
       dens=dens+P(k,1)*((2*a(1)/pi)**0.75)&
          *((2*a(k)/pi)**0.75) *dexp(-(a(1)+a(k))*r2)
     441 enddo
  221 enddo
  fun=0.00d0
  do 151 m=1, nf
     fun=fun+ncoef(m)*((2.0d0*nbasis(m)/pi)**0.750d0)&
        *dexp((-(r2))*(nbasis(m)))
  151 enddo
  psi(i)=su
  den(i)=dens
  del(i)=dsqrt((del(i-1)+(den(i)-fun)**2)/(npt))
  std=del(i)
  total_density=total_density+w(i)*4*pi*r2*den(i)
  write(11,*) xi(i), psi(i), den(i), fun, del(i)
331 enddo
write(14, *) lambda, std,total_density, Energy
write(*,*) total_density
end
!-----!
|-----|
subroutine basis(expnt, coef)
/_____/
use par
implicit none
real*8 :: expnt(nbas), coef(nbas)
CHARACTER(LEN=4) :: var
```

```
INTEGER*4:: II, N1, NUM, IX, IY
N1 = 0
OPEN(11,FILE="coemd-ref")
DO
 READ(11,*) VAR, NUM
 IF ( VAR(1:4) .EQ. "STOP") GOTO 100
   DO IX = 1, NUM
      N1 = N1 + 1
      READ(11, '(I3, E28.16, F21.7)') II, EXPNT(N1), COEF(N1)
   ENDDO
ENDDO
CLOSE(11)
100
    RETURN
end
|-----|
/_____/
subroutine check_constraint(a, P, n, xmin, xmax, rho0, w, r, rho)
/_____/
USE PAR
implicit none
integer(kind=4):: i, j, k, n
real(kind=8) :: r(n), add, rho, rho0, fy, check
real(kind=8) :: xmin, xmax, w(n), a(nbas), P(nbas, nbas)
real(kind=8), parameter:: pi=3.14159265359d0, piby4=pi*0.2500d0
call LEGENDRE(n,xmin, xmax, w, r)
rho=0.00d0
rho0=0.00d0
do 111 i=1,n
   add=0.00d0
   do 221 j=1,nbas
     do 222 k=1, nbas
```

```
add=add+P(j,k)*dexp(-(a(j)+a(k))*r(i)*r(i))
     222 enddo
   221 enddo
   fy=dexp(-(pi/(4.00d0))*r(i)*r(i))
   rho=rho+w(i)*add
   rho0=rho0+w(i)*fy
   check=check+w(i)*(fy-add)*(fy-add)
111 enddo
end
!-----!
subroutine guess_Density_matrix(H, P)
|-----|
USE PAR
implicit none
real(kind=8) :: H(nbas,nbas), P(nbas,nbas), v(nbas,nbas), coef1(nbas,nbas)
real(kind=8) :: E(nbas), a(nbas), c(nbas), coef(nbas,nbas)
integer(kind=4):: i, j, k
             :: ft(nbas,nbas), ft1(nbas,nbas), X(nbas,nbas)
real(kind=8)
call X_matrix(a,c,X)
ft=matmul(H,X)
ft1=matmul(transpose(X),ft)
call CALL_DSYEV_WT_EIGVECS(ft1, nbas, E, coef)
do i=1,nbas
   do j=1,nbas
     P(i,j)=0.0d0
     do k=1, nele/2
        P(i,j)=P(i,j)+2.0d0*coef(i,k)*coef(j,k)
     enddo
   enddo
enddo
end
```

```
|-----|
subroutine rho0(a, c, coef, basis, V0)
|-----|
USE PAR
integer(kind=4):: i, j, k, coef_read
real(kind=8) :: V0(nbas,nbas), a(nbas), c(nbas), S(nbas,nbas), mul
real(kind=8), parameter:: pi=3.14159265359d0, piby4=pi*0.2500d0
real(kind=8) :: s1, s2, s4, alpha, coef(nbas), basis(nbas)
integer(kind=8), parameter:: nf=17
call overlap(a, c, S)
open(15, file='he_density_basis.dat')
do 100 coef_read=1, nf
read(15,*) coef(coef_read), basis(coef_read)
100 enddo
do 111 i=1, nbas
  do 112 j=1, nbas
     s1=a(i)+a(j)
     mul=0.0d0
     do 113 k=1, nf
       s2=(basis(k))
       s4=(s1+s2)
       alpha=2.0d0*dsqrt((s1*s2)/(pi*s4))
       mul=mul+coef(k)*((2.0d0*a(i)/pi)**0.75)*((2*a(j)/pi)**0.75) &
     *((2.0d0*basis(k)/pi)**0.75)*2.00d0*((pi)**2.5)/(s1*s2*dsqrt(s4))
     113 enddo
     VO(i,j)=mul
     write(*,'(7F14.7)') VO(i,j)
   112 enddo
111 enddo
end
!-----!
```

```
subroutine X_matrix(a, c, X)
/-----!
USE PAR
implicit none
integer(kind=4):: i, j, k,l
real(kind=8) :: S(nbas,nbas), a(nbas), c(nbas), s05(nbas,nbas)
real(kind=8) :: st(nbas), U(nbas,nbas)
real(kind=8) :: Ut(nbas,nbas), X(nbas,nbas), su
call overlap(a, c, S)
call CALL_DSYEV_WT_EIGVECS(S, nbas, st, U)
s05=0.0000d0
do 111 i=1, nbas
  s05(i,i)=st(i)**(-0.50d0)
111 enddo
Ut=matmul(s05, transpose(U))
X= matmul(U,Ut)
end
!-----!
subroutine Two_electron_integral(a, c, TEI)
1------
USE PAR
implicit none
integer(kind=4):: i, j, k, l
real(kind=8) :: a(nbas),S(nbas,nbas),c(nbas),
real(kin=8) :: alpha,TEI(nbas,nbas,nbas,nbas)
real(kind=8), parameter:: pi=3.14159265359d0
call overlap(a, c, S)
do 111 i=1,nbas
  do 112 j=1,nbas
     do 113 k=1,nbas
       do 114 l=1,nbas
```

```
alpha=2.0d0*((a(i)+a(j))*(a(k)+a(l))/(pi*(a(i)+a(j) &
               +a(k)+a(1))))**0.50d0
         TEI(i,j,k,l)=S(i,j)*S(k,l)*alpha
       114 enddo
     113 enddo
  112 enddo
111 enddo
end
|-----|
subroutine Hcore(a, c, H)
|-----|
USE PAR
implicit none
real(kind=8) :: V(nbas,nbas)
integer(kind=4):: i, j
real(kind=8) :: omg, H(nbas,nbas)
real(kind=8), parameter:: pi=3.14159265359d0
real(kind=8) :: S(nbas,nbas), a(nbas), c(nbas), T(nbas,nbas)
call Kinetic_energy_matrix(a, c, T)
call overlap(a, c, S)
open(2000,file='test.dat')
do 111 i=1,nbas
  do 112 j=1,nbas
     omg=2*(((a(i)+a(j))/pi)**0.50d0)
     V(i,j)=-nele*omg*S(i,j)
ļ
    V(i,j)=0.0d0 !without (Z/r) term
     H(i,j)=V(i,j)+T(i,j)
  112 enddo
111 enddo
end
!-----!
```
```
subroutine Kinetic_energy_matrix(a, c, T)
/-----/
USE PAR
implicit none
real(kind=8) :: S(nbas,nbas), K(nbas,nbas), T(nbas,nbas)
real(kind=8) :: a(nbas), c(nbas), su
integer(kind=4):: i, j, l
call overlap(a, c, S)
do 111 i=1, nbas
  do 112 j=1, nbas
     su=0.0000d0
     K(i,j)=(3.0d0*a(j)*a(i))/(a(j)+a(i))
     su=K(i,j)*S(i,j)
     T(i,j)=su
   112 enddo
111 enddo
end
|-----|
subroutine overlap(a, c, S)
!-----!
USE PAR
implicit none
real(kind=8), parameter:: xmin=-2.000d0, xmax=2.00d0, pi=3.1415926535
integer(kind=4):: i, j, k, l, p, q, r, k1, k2
integer(kind=4), parameter:: n=1001
real(kind=8) :: x(n), y(n), z(n)
real(kind=8) :: S(nbas,nbas), KE(nbas,nbas), T(nbas,nbas)
real(kind=8) :: a(nbas), su, c(nbas), su1
do 1000 i=1, nbas
  do 1001 j=1, nbas
     S(i,j)=c(i)*c(j)*((4*a(i)*a(j))**0.75d0)/(a(i)+a(j))**1.50d0
```

1000 enddo end	1001 enddo
end	1000 enddo
	end
	!!

Note: The LAPACK libraries are used for diagonalization.

Appendix C

Fitting Code

```
!-----!
Program fitting_3d_function
!-----!
implicit none
integer(kind=4), parameter :: npt=990, nbas=17
integer(kind=4):: i, j, k, info, l
real(kind=8) :: f(npt), c(nbas), a(nbas)
real(kind=8) :: xmin, xmax
real(kind=8) :: x(npt), su, coef(nbas, nbas)
integer(kind=4):: ipiv(nbas)
real(kind=8) :: a2(nbas,nbas), a2inv(nbas,nbas)
real(kind=8) :: b(nbas,1), work(nbas), w(npt)
real(kind=8) :: new_func(npt), fun, S(nbas,nbas)
real(kind=8) :: r(npt), d(nbas,1), integ
         :: norm(1,1), check(1,nbas)
real*8
real*8, parameter:: pi=3.141592653590d0
            :: rmin, rmax, sigma, wr(npt), f1(npt)
real*8
           :: amin,amax, var, var1
real*8
!----Dummies
real*8 :: dum1, dum2, dum4, fx
```

```
open(31,file='he_density.dat') !data file or function file
open(51, file='function.dat')
open(1, file='he_density_basis.dat') !basis set+ coefficients to written
!----reading function
do 111 i=1, npt
   read(31,*) dum1, r(i), dum2, dum4, f(i)
   read(31,*) r(i), dum1, f(i)
   write(51,*) r(i), f(i)
111 enddo
xmin=r(1)
xmax=r(npt)
open(12, file='new_function2.dat')
call basis(nbas, a, coef)
call LEGENDRE(npt ,xmin, xmax, w, x)
!----function/data direct feeding/testing
! do 111 i=1,npt
      f(i)=((dexp(-x(i)/0.529)*(x(i)/(2.0d0*0.5290d0)))\&
!
ļ
            *(2.0d0/0.5290d0)**1.50d0)**2
!111 enddo
!----overlap integrals
call overlap(nbas, a, c, S)
a2inv=S
call DGETRF(nbas, nbas, a2inv, nbas, ipiv, info)
if (info /= 0) then
    write(*,*)'matrix is sigular'
    stop
endif
call DGETRI(nbas, a2inv, nbas, ipiv,work, nbas, info)
if (info /= 0) then
    stop
endif
```

```
!----numerical integrals
do 113 j=1, nbas
   integ=0.00d0
   fun=0.00d0
   do 112 i=1, npt
      fun=fun+w(i)*x(i)*x(i)*4*pi*f(i)*((2.00d0*a(j)/pi)**0.7500d0)&
               *dexp(-a(j)*x(i)*x(i))
   112 enddo
   b(j,1)=fun
113 enddo
do i=1, npt
   integ=integ+w(i)*f(i)*4*pi*x(i)*x(i) !(num)integration of given function
enddo
do i=1,nbas
   su=0.00d0
   do j=1,nbas
      su=su+S(i,j)
   enddo
enddo
d=matmul(a2inv,b)
check=matmul(transpose(d),S)
norm=matmul(check, d)
var=0.00d0
do 121 i=1,npt
    su=0.00d0
    do 131 j=1,nbas
       su=su+(d(j,1))*((2.00d0*a(j)/pi)**0.7500d0)*dexp(-a(j)*x(i)*x(i))
    131 enddo
    new_func(i)=su
    var=var+(new_func(i)-f(i))**2
121 enddo
```

```
fx=0.00d0
do i=1, npt
   write(12,*) x(i), new_func(i), f(i)
   fx=fx+w(i)*4*pi*x(i)*x(i)*new_func(i)
enddo
close(12)
write(*,*) integ, fx, var**0.500d0
!----writing new function in file as summation of basis
do 1 i=1,nbas
   write(1,*) d(i,1), a(i)
1 enddo
end
|-----|
subroutine basis(nbas, expnt, coef)
|-----|
implicit none
CHARACTER(LEN=4):: var
             :: II, N1, NUM, IX, IY, nbas
INTEGER*4
real*8
             :: expnt(nbas), coef(nbas)
N1 = 0
OPEN(11,FILE="coemd-ref")
DO
 READ(11,*) VAR, NUM
IF ( VAR(1:4) .EQ. "STOP") GOTO 100
   DO IX = 1, NUM
     N1 = N1 + 1
     READ(11, '(I3, E28.16, F21.7)') II, EXPNT(N1), COEF(N1)
   ENDDO
ENDDO
CLOSE(11)
100
    RETURN
```

```
end
|-----|
subroutine overlap(nbas, a, c, S)
1------1
implicit none
real(kind=8), parameter:: xmin=-2.000d0, xmax=2.00d0, pi=3.1415926535
integer(kind=4):: i, j, k, l, p, q, r, k1, k2
integer(kind=4):: nbas
real(kind=8) :: S(nbas,nbas)
real(kind=8) :: a(nbas), su, c(nbas), su1
do 1000 i=1, nbas
  do 1001 j=1, nbas
    S(i,j)=((4.00d0*a(i)*a(j))**0.750d0)/(a(i)+a(j))**1.50d0
  1001 enddo
1000 enddo
end
|-----|
```

Note: The LAPACK libraries are used for calculating inverse